

Synthesis and Biological Evaluation of New Analogues of the Active Fungal Metabolites *N*-(2-Methyl-3-oxodecanoyl)-2-pyrroline and *N*-(2-Methyl-3-oxodec-8-enoyl)-2-pyrroline (II)

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New analogues of the bioactive enamides isolated from *P. brevicompactum* (**2** and **3**) have been synthesized to improve the biological activities. Two different structural modifications have been introduced: substitution of the aliphatic side chain present in the natural products (**1–4**) by other groups frequently found in other active compounds and use of other nitrogen-containing five-membered rings with different degrees of oxidation. In this way, the insecticidal and fungicidal activities have been improved. Thus, compound **9**, which possesses a 3-pyrroline ring, exhibited important insecticidal activity against third-instar nymphs of *Oncopeltus fasciatus* Dallas (100% mortality at 7.5 $\mu\text{g}/\text{cm}^2$). Remarkable fungicidal activity was also found, and preliminary structure–activity relationships could be established.

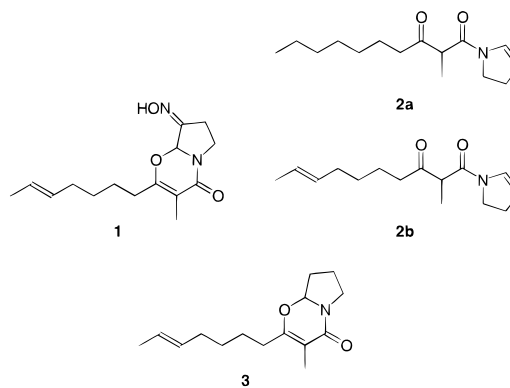
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INTRODUCTION

The synthesis of bioactive natural products is a powerful tool to confirm the structures and activities associated with metabolites which are usually isolated in minimal quantities. This kind of work also leads to a series of potentially active synthetic intermediates, chemically related to the natural compounds. Thus, active natural products can be used as lead molecules to obtain different analogues with common substructures and/or functionalities, sometimes with enhanced activities as compared to those of the reference compounds.

Recently, we have reported on the isolation and identification of bioactive metabolites of fungal origin. The study of the culture broth of *Penicillium brevicompactum* Dierckx led to the isolation and identification of a new family of compounds with important biological activities. One of the most interesting compounds, brevioxime (**1**), exhibits a very high activity as a juvenile hormone (JH) biosynthesis inhibitor (Moya et al., 1997; Castillo et al., 1998). Other metabolites possessing in vivo anti-JH activity, *N*-(2-methyl-3-oxodecanoyl)-2-pyrroline (**2a**) and *N*-(2-methyl-3-oxodec-8-enoyl)-2-pyrroline (**2b**), and insecticidal activity, 2-hept-5-enyl-3-methyl-4-oxo-6,7,8,8a,-tetrahydro-4*H*-pyrrolo[2,1-*b*]-1,3-oxazine (**3**), were synthesized following a common route, which diverged only in the last step; this confirmed the chemical and suggested the biosynthetic relationship between both natural products (**1–3**) (Moya et al., 1998; Cantín et al., 1998).

Owing to the importance of the biological activities of some members of this family and because the



structures were simple enough to warrant consideration as a starting point for synthetic modification, we designed a program aimed at producing analogues with improved biological activity.

The first approach was to use two isolated pyrrolic metabolites, which did not show activity, as lead molecules to obtain related compounds with fungicidal and insecticidal activities (Cantín et al., 1998). More recently, we have reported on a new series of analogues derived from already active enamides with improved activities as compared to those of the natural products (Moya et al., 1999).

As an extension of this work, we wish now to report the synthesis and biological activities of a new series of analogues, where the structural modifications involve important deviations from the parent compounds.

MATERIALS AND METHODS

All chemicals were obtained from commercial suppliers and used without further purification. IR spectra were obtained as liquid films; ν_{max} is given for the main absorption bands. ^1H and ^{13}C NMR spectra were recorded at 300 and 75 MHz, respectively, in CDCl_3 solvent; chemical shifts are reported in

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δ (parts per million) values, using TMS as internal standard. The assignment of ^{13}C signals is supported by DEPT experiments. Mass spectra were obtained under electron impact or chemical ionization; the ratios m/z and the relative intensities are reported. Isolation and purification were done by flash column chromatography on silica gel 60 (230–400 mesh). Analytical TLC was carried out on precoated plates (silica gel 60 F₂₅₄); spots were visualized with UV light and in an I₂ chamber.

General Synthetic Procedures. *Synthesis of β -Oxoamides.* The following procedure was employed with different acyl side chains: To a cooled solution (0 °C) of 2,2-dimethyl-1,3-dioxane-4,6-dione (1.1 mmol) in dichloromethane (1.5 mL) were added pyridine (2.2 mmol) and the corresponding acyl chloride (0.9 mmol) via syringe, dropwise, under nitrogen. The solution was stirred at 0 °C for 1 h, after which time it was allowed to warm to room temperature for an additional period of 2 h. The dichloromethane solution was washed with dilute HCl, water, and brine, dried, and concentrated to dryness to give almost pure the acylated Meldrum's acid, which was used for the aminolysis without further purification.

The acylated Meldrum's acid and pyrrolidine (2.1 mmol) were refluxed in benzene (9.0 mL) for 14 h. The solvent was evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel, to afford the β -oxoamide.

N-(3-Cyclopropyl-3-oxopropanoyl)pyrrolidine (**4c**): 19% yield; obtained as an oil; HRMS (EI), m/z 181.1099 (C₁₀H₁₅NO₂ requires 181.1102); IR ν_{max} 2980, 2960, 2890, 1670, 1600, 1390, 1350, 1310, 1230, 1180, 1170, 1140, 1070, 1035, 990, 970 940, 900, 880, 860, 840, 800, 730, and 720; ^1H NMR δ_{H} 3.6 (s, 2H, H-2'), 3.5 and 3.4 (t + t, J = 7 Hz, 4H, H-2 + H-5), 2.1 (m, 1H, H-1'), 2.0–1.8 (m, 4H, H-3 + H-4), 1.1 and 0.9 (m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_{C} 203.8 (C₃), 164.6 (C_{1'}), 50.4 (C₂), 46.5 (C₂), 45.2 (C₅), 25.3 (C₃), 23.7 (C₄), 20.1 (C_{4'}), and 10.8 (C_{2''} + C_{3''}); MS m/z 181 (M⁺, 46), 166 (11), 153 (13), 140 (5), 138 (6), 124 (3), 112 (35), 98 (14), 96 (5), 84 (11), 70 (100), 69 (26), 55 (18), 43 (10), and 41 (10).

Methylation of β -Oxoamides. A solution of β -oxoamide (0.9 mmol) in DMF (3.0 mL) was added dropwise to a suspension of NaH (60% dispersion oil; 1.1 mmol) (prewashed with pentane) in DMF (1.5 mL) at 0 °C, after which the mixture was warmed to room temperature and stirred for 2.5 h. It was then recooled to 0 °C and treated with iodomethane (1.9 mmol). After being stirred at room temperature for 5.25 h, the mixture was diluted with water and extracted with CH₂Cl₂. The combined extracts were washed with brine, dried, and concentrated to dryness, providing the methylated β -oxoamide.

N-(3-Cyclopropyl-2-methyl-3-oxopropanoyl)pyrrolidine (**5c**): 82% yield; obtained as an oil; HRMS (EI), m/z 195.1268 (C₁₁H₁₇NO₂ requires 195.1259); IR ν_{max} 2960, 2860, 1700, 1630, 1420, 1380, 1330, 1300, 1250, 1220, 1190, 1160, 1130, 1100, 1040, 1010, 940, 910, 870, and 810; ^1H NMR δ_{H} 3.7 (q, J = 7 Hz, 1H, H-2'), 3.5–3.4 (m, 4H, H-2 + H-5), 2.1 (m, 1H, H-1'), 2.0 (m, 4H, H-3 + H-4), 1.4 (d, J = 7 Hz, 3H, CH₃), 1.1–0.9 (m, 4H, H-2'' + H-3''); ^{13}C NMR δ_{C} 207.0 (C₃), 168.0 (C_{1'}), 53.2 (C₂), 46.2 (C₂), 45.6 (C₅), 25.6 (C₃), 23.7 (C₄), 17.8 (C_{1'}), 12.9 (CH₃), 11.1 and 10.9 (C_{2''} + C_{3''}); MS, m/z 195 (M⁺, 52), 180 (5), 167 (23), 152 (4), 138 (6), 127 (64), 126 (63), 110 (8), 99 (12), 98 (31), 84 (7), 70 (100), 69 (61), and 55 (25).

N-(3-Cyclopropyl-2,2-dimethyl-3-oxopropanoyl)pyrrolidine (**7c**): 79% yield from **4a**; obtained as an oil; HRMS (CI), m/z 210.1491 (M + H⁺, C₁₂H₂₀NO₂ requires 210.1494); IR ν_{max} 2980, 2940, 2860, 1690, 1620, 1460, 1410, 1370, 1330, 1250, 1220, 1170, 1160, 1090, 1050, 1010, 1000, 960, 910, 890, 870, 810, and 720; ^1H NMR δ_{H} 3.5 and 3.2 (t + t, J = 7 Hz, 4H, H-2 + H-5), 2.0 (m, 1H, H-1'), 1.9 (m, 4H, H-3 + H-4), 1.4 (s, 6H, 2 × CH₃), 1.0 and 0.9 (m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_{C} 210.5 (C₃), 170.8 (C_{1'}), 56.2 (C₂), 47.1 (C₂), 46.3 (C₅), 26.4 (C₃), 23.2 (C₄), 22.0 (2 × CH₃), 17.5 (C_{1'}), and 11.4 (C_{2''} + C_{3''}); MS (CI), m/z 210 (M + H⁺, 100), 209 (M⁺, 15), 196 (3), 181 (3), 166 (2), 150 (2), 141 (27), 140 (17), 139 (13), 124 (3), and 111 (4).

N-(2-Methyl-4-(3-phenoxyphenyl)-3-oxopentanoyl)pyrrolidine (**5d**): two diastereomers; combined yield 71%; obtained

as oils; HRMS (EI), m/z 351.1830 (C₂₂H₂₅NO₃ requires 351.1834); IR ν_{max} 3040, 2960, 2910, 2865, 1705, 1625, 1570, 1475, 1430, 1360, 1320, 1240, 1150, 1060, 970, 910, 745, and 680; MS, m/z 351 (M⁺, 75), 295 (1), 280 (1), 224 (4), 197 (23), 181 (6), 167 (7), 154 (39), 127 (89), 103 (10), 98 (100), 91 (18), 77 (10), and 55 (2).

The first eluted diastereomer exhibited the following NMR: ^1H NMR δ_{H} 7.3 (m, 2H, H-3''' + H-5'''), 7.2 (m, 1H, H-5''), 7.1 (tt, J = 8 and 1 Hz, 1H, H-4'''), 7.0–6.8 (m, 5H, H-2'' + H-4'' + H-6'' + H-2''' + H-6'''), 4.0 (q, J = 7 Hz, 1H, H-4'), 3.4 (q, J = 7 Hz, 1H, H-2'), 3.3, 3.2, and 2.7 (m + m + m, 4H, H-2 + H-5), 1.8 (m, 4H, H-3 + H-4), 1.4 and 1.3 (d + d, J = 7 Hz, 6H, 2 × CH₃); ^{13}C NMR δ_{C} 206.8 (C₃), 169.2 (C_{1'}), 157.6, 156.6, 142.7 (C_{1''}, C_{3''}, C_{1'''}), 129.9, 129.8, 123.5, 122.7, 118.9, 118.5, 117.0 (C_{2''}, C_{4''}–C_{6''}, C_{2'''}–C_{6'''}), 50.9 (C₄), 49.8 (C₂), 46.6 (C₂), 45.8 (C₅), 25.8 (C₃), 24.1 (C₄), 18.3 and 13.0 (2 × CH₃).

The second eluted diastereomer exhibited the following NMR: ^1H NMR δ_{H} 7.3 (m, 2H, H-3''' + H-5'''), 7.2 (t, J = 8 Hz, 1H, H-5''), 7.1 (tt, J = 8 and 1 Hz, 1H, H-4'''), 7.1–6.9 (m, 4H, H-2'' + H-6'' + H-2''' + H-6'''), 6.8 (ddd, J = 8, 3, and 1 Hz, 1H, H-4'), 4.0 (q, J = 7 Hz, 1H, H-4'), 3.6 (q, J = 7 Hz, 1H, H-2'), 3.4, 3.0, and 2.9 (m + m + m, 4H, H-2 + H-5), 1.8 (m, 4H, H-3 + H-4), 1.4 and 1.3 (d + d, J = 7 Hz, 6H, 2 × CH₃); ^{13}C NMR δ_{C} 206.8 (C₃), 167.8 (C_{1'}), 157.0, 156.6, 141.9 (C_{1''}, C_{3''}, C_{1'''}), 129.7, 129.6, 123.3, 122.9, 118.7, 118.4, 117.2 (C_{2''}, C_{4''}–C_{6''}, C_{2'''}–C_{6'''}), 51.6 (C₄), 50.0 (C₂), 46.0 (C₂), 45.8 (C₅), 26.0 and 25.8 (C₃), 23.9 (C₄), 18.2 and 13.4 (2 × CH₃).

N-(2,2-Dimethyl-4-(3-phenoxyphenyl)-3-oxopentanoyl)pyrrolidine (**7d**): 11% yield as byproduct together with **5d**; obtained as an oil; HRMS (EI), m/z 365.1992 (C₂₃H₂₇NO₃ requires 365.1991); IR ν_{max} 3040, 2960, 2920, 2860, 1700, 1620, 1570, 1480, 1435, 1400, 1240, 1205, and 690; ^1H NMR δ_{H} 7.3 (m, 2H, H-3''' + H-5'''), 7.2 (t, J = 8 Hz, 1H, H-5''), 7.1 (tt, J = 8 and 1 Hz, 1H, H-4'''), 7.0–6.9 (m, 4H, H-2'' + H-6'' + H-2''' + H-6'''), 6.8 (ddd, J = 8, 3, and 1 Hz, 1H, H-4'), 4.1 (q, J = 9 Hz, 1H, H-4'), 3.4, 3.2, 3.0, and 2.6 (m + m + m + m, 4H, H-2 + H-5), 1.6 (m, 4H, H-3 + H-4), 1.5 (s, 3H, (CH₃)₂C), 1.4 (d, J = 7 Hz, 3H, CHCH₃) and 1.3 [s, 3H, (CH₃)₂C]; ^{13}C NMR δ_{C} 210.4 (C₃), 169.4 (C_{1'}), 157.0, 156.8, 142.6 (C_{1''}, C_{3''}, C_{1'''}), 129.7, 129.6, 123.4, 122.7, 118.8, 118.2, 117.2 (C_{2''}, C_{4''}–C_{6''}, C_{2'''}–C_{6'''}), 57.2 (C₂), 47.4 (C₂), 47.2 (C₄), 46.8 (C₅), 25.9 (C₃), 23.8 (CH₃), 22.9 (C₄), 21.8 and 20.2 (2 × CH₃); MS, m/z 365 (M⁺, 41), 224 (7), 197 (28), 168 (24), 141 (100), 112 (19), 104 (9), 98 (62), 91 (7), 77 (7), and 55 (13).

*Anodic Oxidation of *N*-Acylpyrrolidines.* A solution of amide (1.6 mmol) in methanol (60.0 mL) containing tetrabutylammonium *p*-toluenesulfonate (4.4 mmol) as a supporting electrolyte was placed into an electrolysis cell equipped with carbon electrodes (8.5 cm²). A constant current (20 mA) was passed through the solution. After 4.0 F/mol of electricity were passed, the solvent was evaporated under reduced pressure. Water was added to the residue, and the product was extracted with CH₂Cl₂. The combined organic layer was dried over anhydrous sodium sulfate. Thereafter, the drying agent was removed by filtration, the solvent was evaporated to dryness, and the residue was filtered through silica gel using ethyl acetate as eluent, to eliminate the supporting electrolyte. The solvent was evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel, to afford the methoxylated amide.

2-Methoxy-N-(3-cyclopropyl-2-methyl-3-oxopropanoyl)pyrrolidine (**6c**): two diastereomers; combined yield 40%; obtained as oils.

Spectral data of the first eluted diastereomer **6c1**: HRMS (CI), m/z 226.1441 (M + H⁺, C₁₂H₂₀NO₃ requires 226.1443); IR ν_{max} 2920, 2880, 1690, 1635, 1380, 1155, 1140, 1050, 1000, and 810; ^1H NMR δ_{H} 5.6 and 5.0 (d + d, J = 4 Hz, 1H, H-2), 3.8 (q, J = 7 Hz, 1H, H-2'), 3.7 (m, 2H, H-5), 3.4 and 3.3 (s + s, 3H, OMe), 2.1 (m, 2H, H-3), 1.9 (m, 2H, H-4), 1.8 (m, 1H, H-1'), 1.5 and 1.4 (d + d, J = 7 Hz, 3H, CHCH₃), and 1.0 and 0.9 (m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_{C} 208.5 and 207.2 (C₃), 170.2 and 169.7 (C_{1'}), 88.4 and 87.1 (C₂), 56.4 and 54.0 (C₂), 54.2 and 53.5 (OMe), 46.0 and 45.9 (C₅), 31.4 and 30.7 (C₃), 22.9 and 20.8 (C₄), 18.0 and 17.7 (C_{1'}), 14.1 and 13.2 (CH₃),

11.9, 11.8, 11.7, and 11.5 ($C_{2''} + C_{3''}$); MS (CI), m/z 226 ($M + H^+$, 59), 225 (M^+ , 14), 210 (38), 195 (100), 193 (89), 185 (5), 166 (7), 156 (12), 140 (4), 128 (41), and 125 (23).

Spectral data of the second eluted diastereomer **6c2**: HRMS (EI), m/z 225.1375 ($C_{12}H_{19}NO_3$ requires 225.1365); IR ν_{\max} 2920, 2880, 2820, 1690, 1640, 1380, 1155, 1130, 1050, 990, 930, 910, 855, and 810; 1H NMR δ_H 5.5 and 5.0 (d + d, $J = 4$ Hz, 1H, H-2), 3.7 (m, 3H, H-5 + H-2'), 3.4 and 3.2 (s + s, 3H, OMe), 2.1 (m, 2H, H-3), 1.9 (m, 2H, H-4), 1.8 (m, 1H, H-1''), 1.5 and 1.4 (d + d, $J = 7$ Hz, 3H, $CHCH_3$), and 1.0 and 0.9 (m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_C 207.4 and 206.5 (C_3), 170.5 and 170.4 ($C_{1'}$), 88.8 and 87.5 (C_2), 56.6 and 53.8 (C_2), 53.7 and 51.1 (OMe), 46.0 and 45.9 (C_5), 31.4 and 30.5 (C_3), 22.8 and 21.0 (C_4), 18.3 and 18.1 ($C_{1''}$), 13.9 and 13.8 (CH_3), and 11.6 and 11.1 ($C_{2''} + C_{3''}$); MS, m/z 225 (M^+ , 14), 210 (4), 193 (73), 167 (6), 156 (3), 126 (16), 100 (30), 97 (23), 85 (19), 83 (26), 70 (100), and 55 (13).

Synthesis of Enamides. The corresponding methoxy derivative (0.05 mmol) and silica gel (0.05 mmol) were heated at 150–160 °C in a flask, under reduced pressure and nitrogen atmosphere. After 2.75 h, water was added to the residue, and the slurry was extracted with CH_2Cl_2 . The combined organic layer was dried over anhydrous sodium sulfate. The drying agent was then removed by filtration, the solvent was evaporated to dryness, and the residue was purified by column chromatography on silica gel. Under those conditions enamides were obtained; when the reaction was carried out with β -oxoamides, bicyclic oxazines were also formed.

N-(3-Cyclopropyl-2-methyl-3-oxopropanoyl)-2-pyrroline (**2c**) and 3-Methyl-2-cyclopropyl-4-oxo-6,7,8,8a-tetrahydro-4H-pyrrolo[2,1-*b*]-1,3-oxazine (**3c**). The enamide **2c** was an oil obtained in 13% yield: HRMS (EI), m/z 193.1095 ($C_{11}H_{15}NO_2$ requires 193.1102); IR ν_{\max} 3080, 2980, 2860, 1690, 1630, 1410, 1360, 1040, 1000, 900, and 720; 1H NMR δ_H 7.0 and 6.5 (m, 1H, H-2), 5.3 (m, 1H, H-3), 3.9 (t, $J = 9$ Hz, 2H, H-5), 3.7 (q, $J = 7$ Hz, 1H, H-2'), 2.7 and 2.6 (m, 2H, H-4), 2.0 (m, 1H, H-1'), 1.48 and 1.47 (d + d, $J = 3$ Hz, 3H, CH_3), 1.1 and 0.9 (m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_C 207.3 (C_3), 165.5 ($C_{1'}$), 129.4 and 128.3 (C_2), 112.9 and 111.5 (C_3), 54.0 (C_2), 45.5 (C_5), 28.1 (C_4), 18.0 ($C_{1''}$), 13.2 (CH_3), 11.8 and 11.6 ($C_{2''} + C_{3''}$); MS, m/z 193 (M^+ , 20), 125 (3), 124 (2), 96 (7), 69 (100), 68 (47), 55 (7), 53 (4), and 41 (50).

The oxazine **3c** was obtained in 36% yield as a yellow oil: HRMS (EI), m/z 193.1104 ($C_{11}H_{15}NO_2$ requires 193.1102); IR ν_{\max} 3070, 2960, 2920, 2860, 1700, 1640, 1430, 1350, 1250, 1180, 1080, 950, 920, 900, 880, 810, 760, and 640; 1H NMR δ_H 5.1 (dd, $J = 8$ and 4 Hz, 1H, H-8a), 3.7 and 3.4 (m + m, 2H, H-6), 2.4–1.6 (m, 5H, H-7 + H-8 + H-1'), 1.9 (s, 3H, CH_3), and 1.1, 0.9, and 0.7 (m + m + m, 4H, H-2'' + H-3''); ^{13}C NMR δ_C 166.4 (C_4), 163.1 (C_2), 105.6 (C_3), 87.6 (C_{8a}), 44.3 (C_6), 31.6 (C_8), 21.9 (C_7), 10.5 ($C_{1''}$), 9.7 ($C_{1'}$), and 7.8 and 4.4 ($C_2' + C_3'$); MS, m/z 193 (M^+ , 72), 165 (7), 156 (9), 142 (10), 100 (23), 97 (12), 83 (32), 70 (100), 69 (74), and 55 (10).

N-Octanoyl-3-pyrroline (**9**). To a mixture of 3-pyrroline (14.1 mmol) with 1.7 M KOH (9.0 mL) was added a solution of octanoyl chloride (14.0 mmol) in CH_2Cl_2 (9.0 mL) dropwise (10 min). After being stirred at room temperature for 5.5 h, the mixture was extracted with CH_2Cl_2 ; the resulting organic extracts were washed with brine, dried over Na_2SO_4 , and concentrated to dryness to give the *N*-octanoyl-3-pyrroline in a straightforward manner as an oil: 85% yield; HRMS (EI), m/z 195.1619 ($C_{12}H_{21}NO$ requires 195.1623); IR ν_{\max} 3020, 2900, 2820, 1710, 1630, 1605, 1440, 1345, 1320, 1260, 1190, 1100, 1060, 990, 940, 910, 800, 730, 710, and 660; 1H NMR δ_H 5.9 (m, 2H, H-3 + H-4), 4.2 (m, 4H, H-2 + H-5), 2.3 (t, $J = 8$ Hz, 2H, H-2'), 1.7 (m, 2H, H-3'), 1.3 [m, 8H, $(CH_2)_4CH_3$], and 0.9 (t, $J = 6$ Hz, 3H, CH_3); ^{13}C NMR δ_C 171.6 ($C_{1'}$), 126.3 (C_3), 124.8 (C_4), 53.2 (C_2), 53.0 (C_5), 34.3 (C_2), 31.6, 29.3, 29.0, 24.7, 22.4 (C_3-C_7), and 14.0 (CH_3); MS, m/z 195 (M^+ , 75), 180 (1), 169 (6), 166 (6), 153 (22), 143 (3), 138 (12), 127 (11), 124 (29), 111 (100), 110 (57), 96 (19), 84 (19), 69 (92), 68 (99), 57 (39), and 55 (15).

5-[1-(2,5-Dihydro-1H-pyrrolyl)octylidene]-2,2-dimethyl-1,3-dioxane-4,6-dione (**8**). **8** was obtained in 28% yield from octanoyl chloride: obtained as an oil; HRMS (EI), m/z 321.1950

($C_{18}H_{27}NO_4$ requires 321.1940); IR ν_{\max} 3080, 2920, 2840, 1740, 1705, 1660, 1640, 1410, 1390, 1330, 1290, 1150, 1040, 950, 930, 790, 730, and 660; 1H NMR δ_H 6.0 and 5.9 (m + m, 2H, $CH=CH$), 4.6 and 4.5 (br s + br s, 4H, 2 \times NCH_2), 3.2 (t, $J = 8$ Hz, 2H, H-2'), 1.7 [s, 6H, $C(CH_3)_2$], 1.6 (m, 2H, H-3'), 1.3 [m, 8H, $(CH_2)_4CH_3$], and 0.9 (t, $J = 7$ Hz, 3H, CH_3); ^{13}C NMR δ_C 180.3 ($C_{1'}$), 162.2 ($C_4 + C_6$), 126.1 and 122.3 ($CH=CH$), 101.8 (C_2), 82.0 (C_5), 60.7 and 57.8 (2 \times NCH_2), 34.9, 31.4, 29.5, 28.7, 26.9, 26.0, 22.4 [C_3-C_9 , $C(CH_3)_2$], and 13.9 (CH_3); MS, m/z 321 (M^+ , 2), 320 (9), 319 (8), 305 (6), 265 (30), 246 (16), 238 (18), 235 (17), 220 (10), 203 (10), 191 (52), 177 (100), 160 (69), 148 (50), 134 (62), 121 (45), 118 (45), 106 (60), 92 (38), and 81 (46). Anal. Calcd for $C_{18}H_{27}NO_4$: C, 65.75; H, 8.49. Found: C, 65.42; H, 8.58.

N-(3-Oxodecanoyl)-2-pyrrolidinone (**10**) was obtained as an oil following the previously described procedure for β -oxoamides in 70% yield: HRMS (EI), m/z 253.1678 ($C_{14}H_{23}NO_3$ requires 253.1677); IR ν_{\max} 2910, 2840, 1735, 1690, 1610, 1450, 1400, 1360, 1320, 1190, 1160, 1070, 1010, 930, 880, 830, 800, and 720; 1H NMR δ_H 4.0 (s, 2H, H-2'), 3.9 (t, $J = 7$ Hz, 2H, H-5), 2.6 (m, 4H, H-3 + H-4'), 2.1 (m, 2H, H-4), 1.6 (m, 2H, H-5'), 1.3 [br s, 8H, $(CH_2)_4CH_3$], and 0.9 (t, $J = 7$ Hz, 3H, CH_2CH_3); ^{13}C NMR δ_C 203.7 (C_3), 175.6 (C_2), 167.2 ($C_{1'}$), 51.4 (C_2), 45.1 (C_5), 42.9 (C_4), 33.8, 33.1, 31.5, 28.9, 23.2, 22.5 (C_3 , C_5-C_9), 16.8 (C_4), and 13.9 (CH_3); MS, m/z 253 (M^+ , 6), 235 (1), 211 (4), 182 (6), 169 (100), 154 (42), 150 (9), 127 (40), 112 (12), 99 (19), 86 (99), 69 (6), and 57 (40).

N-(2-Methyl-3-oxodecanoyl)-2-pyrrolidinone (**11**) was obtained as an oil in 70% yield following the same procedure described previously for β -oxoamides: HRMS (EI), m/z 267.1833 ($C_{15}H_{25}NO_3$ requires 267.1834); IR ν_{\max} 2920, 2850, 1730, 1685, 1450, 1400, 1350, 1240, 1120, 1010, 910, 830, and 715; 1H NMR δ_H 4.5 (q, $J = 7$ Hz, 1H, H-2'), 3.9 (m, 2H, H-5), 2.7 (m, 2H, H-3), 2.6 (t, $J = 8$ Hz, 2H, H-4'), 2.1 (m, 2H, H-4), 1.6 (m, 2H, H-5'), 1.4 (d, $J = 7$ Hz, 3H, $CHCH_3$), 1.3 [br s, 8H, $(CH_2)_4CH_3$], and 0.9 (t, $J = 7$ Hz, 3H, CH_2CH_3); ^{13}C NMR δ_C 207.7 (C_3), 175.6 (C_2), 170.8 ($C_{1'}$), 53.8 (C_2), 45.6 (C_5), 40.8 (C_4), 33.5, 31.6, 29.1, 23.3, 22.5 (C_3 , C_5-C_9), 17.0 (C_4), and 14.0 and 12.6 (2 \times CH_3); MS, m/z 267 (M^+ , 4), 196 (1), 183 (59), 168 (20), 141 (100), 127 (39), 113 (57), 86 (87), 83 (17), 69 (10), 57 (79), and 55 (14).

N-(2-Methyl-3-hydroxydecanoyl)-2-pyrrolidinone (**12**). A solution of the ketoimide **11** (170 mg, 0.6 mmol) in CH_2Cl_2 (35.0 mL) was cooled at -30 °C, and $Zn(BH_4)_2$ (0.14 M in diethyl ether; 4.5 mL, 0.6 mmol) was added to it. After the mixture had been stirred for 1.25 h, getting to -20 °C, it was treated with acetone and then allowed to warm to room temperature. The mixture was diluted with CH_2Cl_2 and washed with water and brine, dried, and concentrated to give an oily residue, which was purified by column chromatography on silica gel using hexane/EtOAc (8:2) as eluent to afford the β -hydroxyimide **12** (82 mg, 49%) as a yellow oil: HRMS (EI), m/z 270.2067 ($M + H^+$, $C_{15}H_{28}NO_3$ requires 270.2069); IR ν_{\max} 3450, 2920, 2840, 1740, 1690, 1450, 1350, 1250, 1220, 1090, 1020, 970, 930, 890, and 830; 1H NMR δ_H 3.9 (m, 1H, H-3'), 3.8 (td, $J = 7$ and 2 Hz, 2H, H-5), 3.7 (m, 1H, H-2'), 3.0 (d, $J = 3$ Hz, 1H, OH), 2.6 (t, $J = 8$ Hz, 2H, H-3), 2.0 (m, 2H, H-4), 1.3 [br s, 8H, $(CH_2)_4CH_3$], 1.2 (d, $J = 7$ Hz, 3H, $CHCH_3$), and 0.9 (t, $J = 7$ Hz, 3H, CH_2CH_3); ^{13}C NMR δ_C 178.7 ($C_{1'}$), 175.3 (C_2), 71.5 (C_3), 45.7 (C_5), 42.9 (C_2), 33.9, 33.8, 31.8, 29.6, 29.2, 26.0, 22.6 (C_3 , C_4-C_9), 17.0 (C_4), and 14.1 and 10.0 (2 \times CH_3); MS, m/z 270 ($M + H^+$, 3), 251 (9), 183 (4), 170 (27), 166 (13), 141 (100), 113 (59), 98 (13), 86 (99), 69 (10), 57 (14), and 55 (14).

Biological Activity. *Insects.* *Oncopeltus fasciatus* Dallas were maintained at 28 ± 1 °C, 50–60% relative humidity, 16 h/8 h (light/dark) photoperiod, and on a diet based on sunflower seeds.

Target Microorganisms. Fungicidal activity was measured against 13 agronomically important phytopathogens: *Aspergillus parasiticus* (CECT 2681), *Geotrichum candidum* (CCM 245), *Alternaria tenuis* (CECT 2662), *Colletotrichum gloeosporioides* (CECT 2859), *Colletotrichum coccodes* (CCM 327), *Fusarium oxysporum* ssp. *gladioli* (CCM 233), *Fusarium oxysporum* ssp. *niveum* (CCM 259), *Fusarium culmorum* (CCM 172), *Penicillium italicum* (CECT 2294), *Trichoderma viride*

(CECT 2423), *Trichothecium roseum* (CECT 2410), *Rosellinia necatrix* (CCM 297), and *Verticillium dahliae* (CCM 269).

The strains were provided by the Colección Española de Cultivos Tipo (CECT) or by the Colección de la Cátedra de Microbiología (CCM) of the Department of Biotechnology (Universidad Politécnica de Valencia).

Entomotoxicity and Anti-JH Activity. The test was carried out basically according to the contact method of Bowers et al. (1976). Briefly, 15 third-instar *O. fasciatus* nymphs were confined to a 9 cm Petri dish coated, across the bottom, with 20 $\mu\text{g}/\text{cm}^2$ of the product. Toxicity effects were considered according to the number of insects dead after 72 h of exposure to the chemicals. The surviving nymphs were transferred to a 500 cm^3 glass flask and held at standard conditions. After metamorphosis occurred and reproduction was successful with the production of viable offsprings, the tests were finished. The tests were considered to be positive for JH antagonistic activity when precocious metamorphosis occurred. Controls were run in parallel and received the same amount of acetone as treated insects.

Antifungal Activity. The products, dissolved in acetone, were added to PDA, in a concentration 100 $\mu\text{g}/\text{mL}$. PDA plates containing only acetone were used as control plates, and a positive control with benomyl [methyl-1-(butylcarbamoyl)-2-benzimidazolecarbamate; Sigma] at 2.5 $\mu\text{g}/\text{mL}$ was performed to appraise the level of activity of the synthesized compounds. Spores from 7-day-old cultures of each fungus on PDA plates were used as an inoculum onto the control and test plates. The radial mycelial growth was measured, and the percentage of inhibition was calculated on the basis of growth in control plates, after 4 days of incubation (6 days for *R. necatrix* and *V. dahliae*), at 28 °C. The antifungal activity of each product was determined three times. Analysis of variance (ANOVA) was performed for fungicidal data (Table 1), and the least significant difference (LSD) test was used to compare means (Statgraphics Plus 4.0).

RESULTS AND DISCUSSION

Recently, we have reported on the chemical synthesis of some biologically active natural products previously isolated in our laboratories (**2a**, **2b**, and **3**). In the course of these studies we obtained a series of intermediates that showed interesting activities, improving in some cases the activities found for the natural products.

These results encouraged us to introduce certain modifications in the synthetic sequences that could lead to related active analogues; the ultimate goal would be to improve the activities of the natural products. Thus, we decided to introduce two different changes: (a) substitution of an aliphatic side chain present in the natural products by groups frequently found in other active compounds and (b) use of other nitrogen-containing five-membered rings with different degrees of oxidation.

In connection with the first approach, some functional groups present in known active compounds were considered taking into account their compatibility with the required reaction conditions and the availability of the corresponding reagents. Hence, cyclopropyl (present in synthetic pyrethroids) and phenoxyphenyl (very common in pesticides) were selected for this work.

The chlorides of cyclopropanecarboxylic acid and fenoprofen [2-(3-phenoxyphenyl)propionic acid] were used as starting materials (Scheme 1).

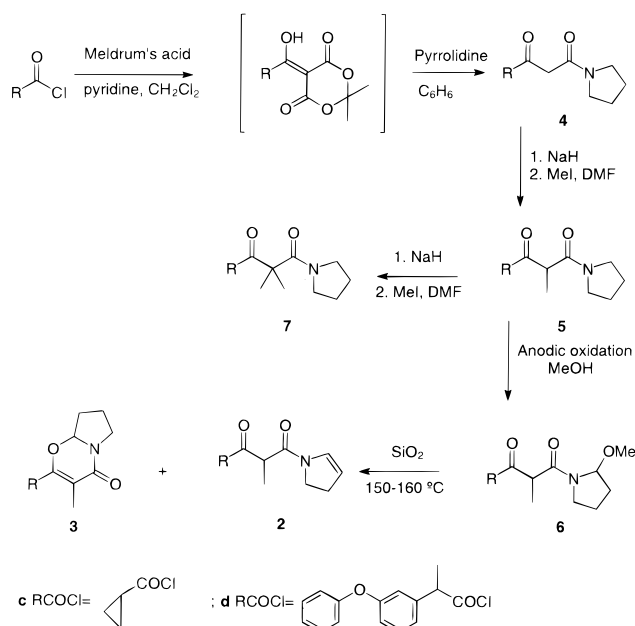
Briefly summarized, the reaction sequence implied formation of acylated Meldrum's acid (Meldrum, 1908; Davidson and Bernhardt, 1948; Oikawa et al., 1978) as first step in the construction of the enamide ring. Subsequent aminolysis (Pak et al., 1992) with pyrrolidine and alkylation (Benetti and Romagnoli, 1995; Abad

Table 1. Analogues Showing Fungicidal Activity

product	radial mycelial growth inhibition, % (mean \pm SD) ^a												
	1	2	3	4	5	6	7	8	9	10	11	12	13
3c	38.9 \pm 0.7 ^A	24.1 \pm 0.9 ^A	26.2 \pm 1.6 ^A	0	11.8 \pm 0.3 ^A	37.8 \pm 2.0 ^A	0	23.9 \pm 3.0 ^{AE}	65.4 \pm 1.2 ^A	37.5 \pm 3.3 ^A	0	23.6 \pm 4.2 ^A	11.3 \pm 1.6 ^A
5d	23.5 \pm 3.9 ^{CB}	23.1 \pm 1.6 ^A	17.4 \pm 2.7 ^B	12.3 \pm 2.5 ^A	41.4 \pm 1.3 ^B	37.6 \pm 1.5 ^A	46.6 \pm 3.4 ^A	39.5 \pm 0.8 ^B	19.0 \pm 3.3 ^B	25.9 \pm 4.2 ^B	30.7 \pm 1.2 ^A	22.2 \pm 3.9 ^{AB}	27.9 \pm 2.7 ^B
7d	16.3 \pm 3.7 ^D	34.8 \pm 3.2 ^{BC}	36.9 \pm 2.7 ^C	11.7 \pm 2.9 ^A	62.2 \pm 5.9 ^C	55.7 \pm 1.8 ^B	53.2 \pm 2.0 ^B	55.2 \pm 3.3 ^C	22.7 \pm 2.8 ^B	48.5 \pm 1.3 ^C	38.7 \pm 3.1 ^B	17.8 \pm 3.9 ^B	37.9 \pm 2.3 ^C
8	39.7 \pm 2.0 ^A	26.0 \pm 1.8 ^A	28.0 \pm 1.0 ^{AD}	23.9 \pm 0.5 ^B	29.9 \pm 2.6 ^B	18.8 \pm 2.8 ^C	38.6 \pm 3.0 ^C	21.1 \pm 4.1 ^A	13.9 \pm 2.5 ^C	18.3 \pm 2.5 ^D	54.0 \pm 4.0 ^C	29.9 \pm 2.0 ^{CD}	39.3 \pm 1.2 ^C
9	24.5 \pm 4.2 ^C	38.5 \pm 1.8 ^C	30.5 \pm 3.3 ^{DE}	0	35.4 \pm 2.4 ^B	76.7 \pm 5.4 ^D	34.3 \pm 1.7 ^D	51.4 \pm 7.5 ^{CD}	39.6 \pm 3.6 ^D	48.3 \pm 2.9 ^C	38.0 \pm 2.0 ^B	31.2 \pm 3.2 ^{CD}	19.3 \pm 1.2 ^D
10	8.2 \pm 1.0 ^E	35.5 \pm 5.1 ^B	27.6 \pm 1.7 ^{AD}	0	34.5 \pm 3.7 ^B	48.3 \pm 3.5 ^E	0	30.2 \pm 4.3 ^E	49.8 \pm 0.3 ^F	26.7 \pm 2.7 ^B	22.0 \pm 2.0 ^D	26.4 \pm 2.0 ^{AC}	21.3 \pm 2.3 ^D
11	18.1 \pm 1.9 ^{BD}	26.8 \pm 4.7 ^A	21.0 \pm 1.8 ^B	24.4 \pm 0.9 ^B	35.6 \pm 2.3 ^B	52.0 \pm 1.8 ^{BE}	49.5 \pm 1.8 ^{AB}	47.5 \pm 4.9 ^D	56.2 \pm 3.0 ^F	27.6 \pm 2.5 ^B	21.1 \pm 1.8 ^D	8.6 \pm 0.7 ^E	31.7 \pm 1.5 ^E
12	24.3 \pm 4.4 ^C	36.5 \pm 1.7 ^{BC}	33.3 \pm 1.7 ^{EC}	0	36.6 \pm 3.9 ^B	69.0 \pm 3.5 ^F	19.6 \pm 3.4 ^E	38.1 \pm 1.7 ^B	45.9 \pm 3.6 ^F	36.7 \pm 2.9 ^A	38.0 \pm 3.5 ^B	33.3 \pm 2.1 ^D	31.5 \pm 1.3 ^E
benomyl	87.0 \pm 1.4	11.1 \pm 0.0	100	0	100	100	100	0	100	100	100	100	100

^a Values represent means \pm standard deviations of growth inhibitions from three independent experiments. Assays concentration of analogues: 100 $\mu\text{g}/\text{mL}$. Benomyl concentration: 2.5 $\mu\text{g}/\text{mL}$. Within each column, mean values showing the same superscripts (A–F) are not significantly different ($P > 0.05$). Target plant pathogens: 1, *F. culmorum*; 2, *F. oxysporum* ssp. *gladioli*; 3, *F. oxysporum* ssp. *niveum*; 4, *G. candidum*; 5, *C. coccoides*; 6, *C. coccoides*; 7, *T. roseum*; 8, *A. tenuis*; 9, *V. dahliae*; 10, *P. citrophthora*; 11, *T. viride*; 12, *P. italicum*; 13, *A. parasiticum*.

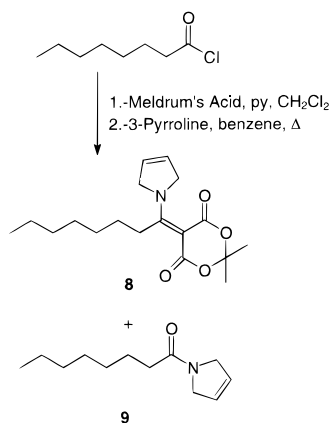
Scheme 1



et al., 1997) provided the β -ketoamide system. In the case of the cyclopropane derivative, anodic oxidation (Shono, 1984; Shono et al., 1981a,b, 1982a,b) followed by elimination of MeOH heating at 150–160 °C (Slomczynska et al., 1996; Cornille et al., 1994, 1995; Moeller et al., 1992, 1994) afforded the β -ketoamide **2c**, along with the bicyclic isomer **3c**.

However, when fenoprofen was used as starting product, the anodic oxidation step led to a complex reaction mixture including some products arising from oxidation of the phenoxyphenyl group; in view of this result, no attempts were done to isolate the desired methoxylated products.

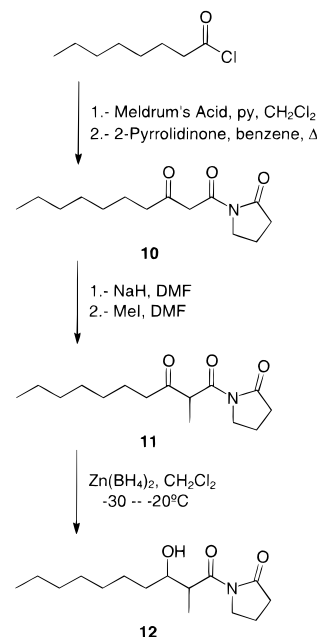
In the second approach, the modifications affected the nature of the five-member ring. Thus, 2-pyrrolidinone and 3-pyrroline were used to carry out the aminolysis of the Meldrum's acid derivative. With 3-pyrroline, the



enamine product **8** was obtained as side product, together with the monocarbonylic amide **9** formed by direct reaction of unreacted octanoyl chloride with 3-pyrroline. The structure of this byproduct was proved by direct synthesis by means of a Schotten–Baumann reaction.

When 2-pyrrolidinone was employed, the expected imide was obtained in good yield; the same was true with the subsequent alkylation of activated position.

Finally, reduction of the ketone group (Evans and DiMare 1986; Evans et al., 1984; Nakata et al., 1982; Nakata and Oishi, 1980; Saksena and Mangiaracina, 1983; Eguchi et al., 1996) was carried out using $\text{Zn}(\text{BH}_4)_2$ (Gensler et al., 1960; Wiberg 1953). Although



cyclization of this compound with formation of a hemiketal could give rise to the heterobicyclic system present in **3**, with a different functionalization, such a process was not observed with compound **12**.

Biological Activities. *Fungicidal Activity.* Table 1 shows the fungicidal activity of the new analogues, expressed as the percentage of growth inhibition against different agronomically important plant pathogens.

At first sight, it is interesting to note that although none of the analogues were strongly effective in the inhibition of the growth of tested microorganisms [comparatively, the levels of activity are clearly lower than those of a conventional fungicide such as benomyl (Table 1)], the data obtained in this paper, together with those recently reported (Moya et al., 1999), allowed us to establish preliminary structure–activity relationships.

Regarding the first approach, compound **7d** possessing the phenoxyphenyl substituent yielded the best fungicidal activity, as it showed growth inhibitions >50% for *C. gloesporoides*, *T. roseum*, and *A. tenuis*; in addition, substantial inhibitions of other five fungal species were also obtained. The second phenoxyphenyl-substituted product (**5d**), although considerably active against *C. gloesporoides* and *T. roseum*, did not exhibit percentages of inhibition >50%. This fact suggests that the double methylation in the β -ketoamide system enhances the fungicidal activity.

On the other hand, introduction of the cyclopropyl group resulted in an adverse effect on the activity; only compound **3c** showed an important activity against *V. dahliae*, which was still remarkable against *F. culmorum*, *C. coccodes*, and *P. citrophthora*.

The second synthetic approach gave higher but more selective growth inhibitions. Compound **9**, possessing 3-pyrroline instead of pyrrolidine, was highly active against *C. coccodes* (~75%), showing moderate activity against the other nine fungi.

Products obtained when pyrrolidine was substituted by 2-pyrrolidinone yielded different levels of activity. The best one, with regard to the spectrum of affected fungi, was found with compound **11**. However, product **12** was particularly active against *C. coccodes*, showing significant differences with the latter compound. Thus, it seems that reduction of the ketone group selectively increases the activity against this fungal species. Finally, the lack of a methyl group between the carbonyls in these structures (compound **10**) produced a decreased fungicidal activity in all cases, suggesting that the methyl group, which likely provides rigidity to the molecule, is important in conferring activity to the products. A possible explanation of the above results could be that **11** and **12** contain modifications that reduce acidity of the α -carbonyl proton; this may be related to metabolic stability. However, at the moment, this is a speculative hypothesis and remains to be proved.

Insecticidal Activity. Only compounds possessing a 3-pyrroline ring showed insecticidal activity. Product **9** was highly active against *O. fasciatus*, exhibiting 100% mortality at a dose of 7.5 $\mu\text{g}/\text{cm}^2$; at lower doses the toxicity decreased considerably, exhibiting 20% mortality at 5.0 $\mu\text{g}/\text{cm}^2$. Compound **8** was less active, showing a percentage of mortality of 40% at a dose of 10 $\mu\text{g}/\text{cm}^2$.

The rest of the compounds did not show activity under our assay conditions.

As mentioned above, important improvements in biological activities have been achieved either in this or in our previous study (Moya et al., 1999), which was based also on the synthesis of analogues using the active pyrroline natural products as starting points. Thus, the reported success of this approach, combined with the growing need to develop new products for ecologically acceptable programs of pest control, makes this kind of work an attractive option for biorational pesticide design.

LITERATURE CITED

- Abad, A.; Agulló, C.; Arnó, M.; Cantín, A.; Cuñat, A. C.; Mesguer, B.; Zaragoza, R. J. Stereoselective synthesis of (-)-metasequoic acid B. *J. Chem. Soc., Perkin Trans. 1* **1997**, 1837–1843.
- Benetti, S.; Romagnoli, R. Mastering β -keto esters. *Chem. Rev.* **1995**, *95*, 1065–1114.
- Cantín, A.; Moya, P.; Miranda, M. A.; Primo, J.; Primo-Yúfera, E. Isolation of *N*-(2-methyl-3-oxodecanoyl)pyrrole and *N*-(2-methyl-3-oxodec-8-enoyl)pyrrole, two new natural products from *Penicillium brevicompactum*, and synthesis of analogues with insecticidal and fungicidal activity. *J. Agric. Food Chem.* **1998**, *46*, 4748–4753.
- Castillo, M.; Moya, P.; Couillaud, F.; Garcerá, M. D.; Martínez-Pardo, R. A heterocyclic oxime from a fungus with anti-juvenile hormone activity. *Arch. Insect. Biochem. Physiol.* **1998**, *37*, 287–294.
- Cornille, F.; Fobian, Y. M.; Slomczynska, U.; Beusen, D. D.; Marshall, G. R.; Moeller, K. D. Anodic amide oxidations: conformationally restricted peptide building blocks from the direct oxidation of dipeptides. *Tetrahedron Lett.* **1994**, *35*, 6889–6992.
- Cornille, F.; Slomczynska, U.; Smythe, M. L.; Beusen, D. D.; Marshall, G. R.; Moeller, K. D. Electrochemical cyclization of dipeptides toward novel bicyclic, reverse-turn peptidomimetics. Synthesis and conformational analysis of 7,5-bicyclic systems. *J. Am. Chem. Soc.* **1995**, *117*, 909–917.
- Davidson, D.; Bernhardt, S. A. The structure of Meldrum's supposed β -lactonic acid. *J. Am. Chem. Soc.* **1948**, *70*, 3426–3428.
- Eguchi, S.; Suzuki, T.; Okawa, T.; Matsushita, Y. Synthesis of optically active Vasicinone based on intramolecular azo-Wittig reaction and asymmetric oxidation. *J. Org. Chem.* **1996**, *61*, 7316–7319.
- Evans, D. A.; DiMare, M. Asymmetric synthesis of Premonensin, a potential intermediate in the biosynthesis of Monensin. *J. Am. Chem. Soc.* **1986**, *108*, 2476–2478.
- Evans, D. A.; Ennis, M. D.; Le, T. Asymmetric acylation reactions of chiral imide enolates. The first direct approach to the construction of chiral β -dicarbonyl synthons. *J. Am. Chem. Soc.* **1984**, *106*, 1154–1156.
- Gensler, W. J.; Johnson, F.; Sullivan, W. F. Compounds related to podophyllotoxin. XI. An unusual Stobbe condensation. *J. Am. Chem. Soc.* **1960**, *82*, 6070–6074.
- Meldrum, A. N. A β -lactonic acid from acetone and malonic acid. *J. Chem. Soc.* **1908**, *93*, 598–601.
- Moeller, K. D.; Rutledge, L. D. Anodic amide oxidations: The synthesis of two spirocyclic L-pyrroglutamide building blocks. *J. Org. Chem.* **1992**, *57*, 6360–6363.
- Moeller, K. D.; Hanau, C. E.; Avignon, A. The use of HMQC-TOCSY experiments for elucidating the structures of bicyclic lactams: uncovering a surprise rearrangement in the synthesis of a key PRO-PHE building block. *Tetrahedron Lett.* **1994**, *35*, 825–828.
- Moya, P.; Castillo, M.; Primo-Yúfera, E.; Couillaud, F.; Martínez-Mañez, R.; Garcerá, M. D.; Miranda, M. A.; Primo, J.; Martínez-Pardo, R. Brevoxime, a new juvenile hormone biosynthesis inhibitor isolated from *Penicillium brevicompactum*. *J. Org. Chem.* **1997**, *62*, 8544–8545.
- Moya, P.; Cantín, A.; Castillo, M. A.; Primo, J.; Miranda, M. A.; Primo-Yúfera, E. Isolation, structural assignment and synthesis of *N*-(2-methyl-3-oxodecanoyl)-2-pyrroline, a new natural product from *Penicillium brevicompactum* with in vivo anti-juvenile hormone activity. *J. Org. Chem.* **1998**, *63*, 8530–8535.
- Moya, P.; Cantín, A.; Miranda, M. A.; Primo, J.; Primo-Yúfera, E. Synthesis and biological activities of new analogues of the active fungal metabolites *N*-(2-methyl-3-oxodecanoyl)-2-pyrroline and *N*-(2-methyl-3-oxodec-8-enoyl)-2-pyrroline. *J. Agric. Food Chem.* **1999**, *47*, 3866–3871.
- Nakata, T.; Oishi, T. Stereoselective reduction of β -keto esters with zinc borohydride. Stereoselective synthesis of erythro-3-hydroxy-2-alkylpropionates. *Tetrahedron Lett.* **1980**, *21*, 1641–1644.
- Nakata, T.; Kuwabara, T.; Tani, Y.; Oishi, T. Total synthesis of (\pm)-oudemansin. *Tetrahedron Lett.* **1982**, *23*, 1015–1016.
- Oikawa, Y.; Sugano, K.; Yonemitsu, O. Meldrum's acid in organic synthesis. A general and versatile synthesis of β -keto esters. *J. Org. Chem.* **1978**, *43*, 2087–2088.
- Pak, C. S.; Yang, H. C.; Choi, E. B. Aminolysis of 5-acyl-2,2-dimethyl-1,3-dioxane-4,6-diones (acyl Meldrum's acids) as a versatile method for the synthesis of β -oxo carboxamides. *Synthesis* **1992**, 1213–1214.
- Saksena, A. K.; Mangiaracina, P. Recent studies on veratrum alkaloids: a new reaction of sodium triacetoxymethylborohydride $[\text{NaBH}(\text{OAc})_3]$. *Tetrahedron Lett.* **1983**, *24*, 273–276.
- Shono, T. Electroorganic chemistry in organic synthesis. *Tetrahedron Lett.* **1984**, *40*, 811–850.
- Shono, T.; Matsumura, Y.; Tsubata, K. Electroorganic chemistry. A new carbon-carbon bond forming reaction at the α -position of amines utilizing anodic oxidation as a key step. *J. Am. Chem. Soc.* **1981a**, *103*, 1172–1176.
- Shono, T.; Matsumura, Y.; Tsubata, K. A new synthetic method of α -amino acids from α -methoxyurethanes. *Tetrahedron Lett.* **1981b**, *22*, 2411–2412.
- Shono, T.; Matsumura, Y.; Tsubata, K.; Sugihara, Y.; Yamane, S.; Kanazawa, T.; Aoki, T. Electroorganic Chemistry. Electroorganic synthesis of enamides and enecarbamates and their utilization in organic synthesis. *J. Am. Chem. Soc.* **1982a**, *104*, 6697–6703.
- Shono, T.; Matsumura, Y.; Tsubata, K.; Sugihara, Y. A new method of acylation at β -position of aliphatic amines. *Tetrahedron Lett.* **1982b**, *23*, 1201–1204.
- Slomczynska, U.; Chalmers, D. K.; Cornille, F.; Smythe, M. L.; Beusen, D. D.; Moeller, K. D.; Marshall, G. R. Electrochemical cyclization of dipeptides to form novel bicyclic, reverse-turn peptidomimetics. Synthesis and conformational

analysis of 6,5-bicyclic systems. *J. Org. Chem.* **1996**, *61*, 1198–1204.

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